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an acid and a base and produces water and a salt. The final solution is closer to a neutral pH than either the acid or base alone. Hydrochloric acid (HCl) reacts with potassium hydroxide (KOH), producing a solution of potassium chloride (KCl) and water (H. 2O).

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**Answers** that has the same properties throughout; a mixture in which one substance is dissolved into another. solvent.

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