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The elements

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carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things. They form the carbohydrates, nucleic acids, proteins, and lipids (all of which will be defined later in this chapter) that are

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the fundamental

molecular  
Chapter 24

components of all organisms. In this chapter, we will discuss these important building blocks and learn how the unique properties of the atoms of different elements affect their interactions with other ...

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The chemistry of  
life begins with the

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basic principles of  
bond formation and  
bond breaking, and  
the nature of the  
different  
compounds  
formed. Life  
revolves around  
the balancing act  
between the  
energy released as  
bonds are broken  
and the energy  
taken in as bonds



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are formed. Life on  
earth depends on  
the nature of the  
carbon atom and  
the nature of  
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For example, in  
Biology Chapter 2

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the Chemistry of  
Life, students learn  
about carbon,

nitrogen, oxygen,  
hydrogen, and  
iodine. The

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teaches students  
the correct and  
incorrect ways of  
using the scales to  
solve problems and  
science units.

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before, the online  
students will not be  
lectured about  
biology.

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the Chemistry Of  
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Answers~~

SAMPLE ANSWER:  
Carbon is the  
primary element  
found in living  
things. SAMPLE

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ANSWER: Answer Key

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Organisms use  
carbon compounds  
to form four types  
of molecules:

lipids,

carbohydrates,

nucleic acids, and  
proteins. SAMPLE

ANSWER: A lot of  
what that happens  
in an organism is  
based on chemical  
reactions. SAMPLE

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ANSWER: Enzymes  
are proteins that  
speed up

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Your job will  
include things like  
quality control

dosimetry, therapy  
planning and other  
tasks that are

essential, but your  
occupation

environment might  
be different, based

on the place you

work. Your capacity

to earn is based on

the amount of work

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—Semesprit~~

The four main types of carbon-based molecules in organisms are carbohydrates, lipids, nucleic acids, and starches. fatty

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monosaccharides.  
proteins. Q. Both  
animal fats and  
plant oils are made  
up of glycerol and  
phospholipids. fatty  
acids.

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matter pages 35

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Section 2-1 The

Nature of Matter

(pages 35-39) This

section identifies

the three particles

that make up

atoms. It also

explains how

atoms of the same

element can have

a different number

of neutrons and

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describes the two  
main types of  
chemical bonds.

Atoms (page 35) 1.

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TESTING FOR

BIOMOLECULES

Data Sheet Table

1. Testing For

Proteins Results

Sample Initial Color

Hypothesis (Protein

+/-) Final Color

Protein Present?

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Albumin (1) Gelatin  
(2) Glucose (3)  
Water (4) E Table  
2.

~~Solved: The  
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EXERCISE 1:  
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Section 2-1 The  
Nature of Matter

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(pages 35–39) This section identifies the three particles that make up atoms. It also explains how atoms of the same element can have a different number of neutrons and describes the two main types of chemical bonds.

Atoms (page 35) 1.

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